

34<sup>th</sup> JSMCWM annual meeting & 3RINCs Autumn

Hydrogen production from Organic waste using reforming reaction











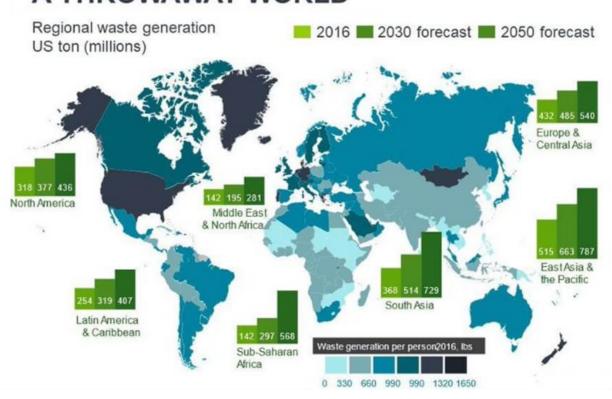






## Waste to energy

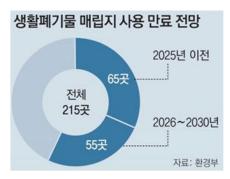
#### A THROWAWAY WORLD



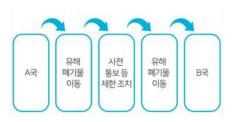
**Increase of waste** 



#### Ocean disposal X



**Lack of landfill** 



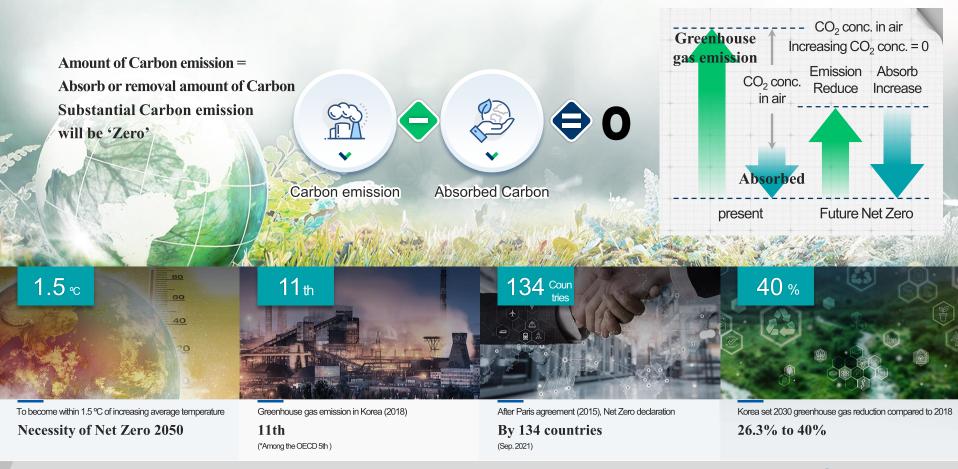
#### **Limitation of waste trade**

폐기물의 기존처리 방법(해양투기, 매립, 소각 등)을 대체할 수 있는 전략 마련의 필요성이 증대됨



# Carbon neutrality, Net Zero

Net zero means that  $CO_2$  emissions due to the human activity are in balance with global absorption, so that the concentration of  $CO_2$  in the atmosphere does not increase further It is necessary to reduce anthropogenic emissions of  $CO_2$  and increase the amount of absorption (removal) using forest restoration and negative emission technology\*

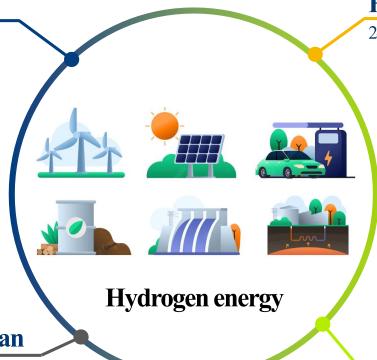




# Hydrogen energy

#### **Alternative energy**

The way to replace fossil fuels



#### High energy efficiency

2~3 times higher than thermal engine

### **Eco-friendly and clean**

Less greenhouse gas emission and pollutants

### **Sustainable energy**

Unlimited amount energy source comparing to fossil fuels

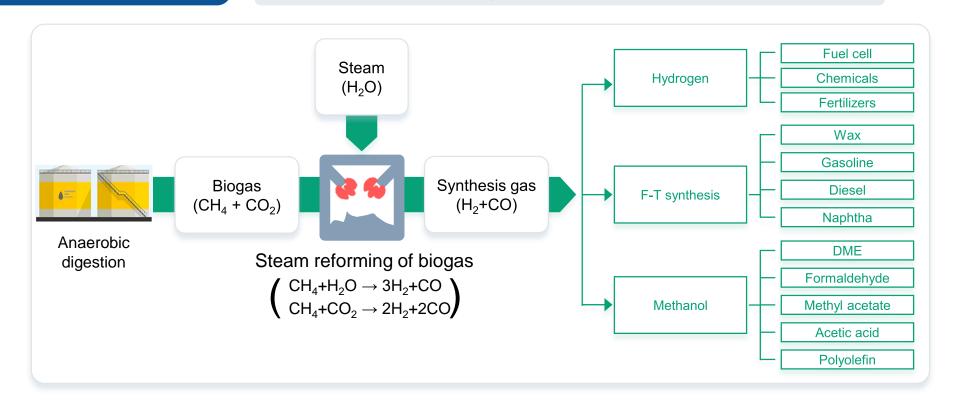


## Organic waste to hydrogen

# Steam reforming of biogas

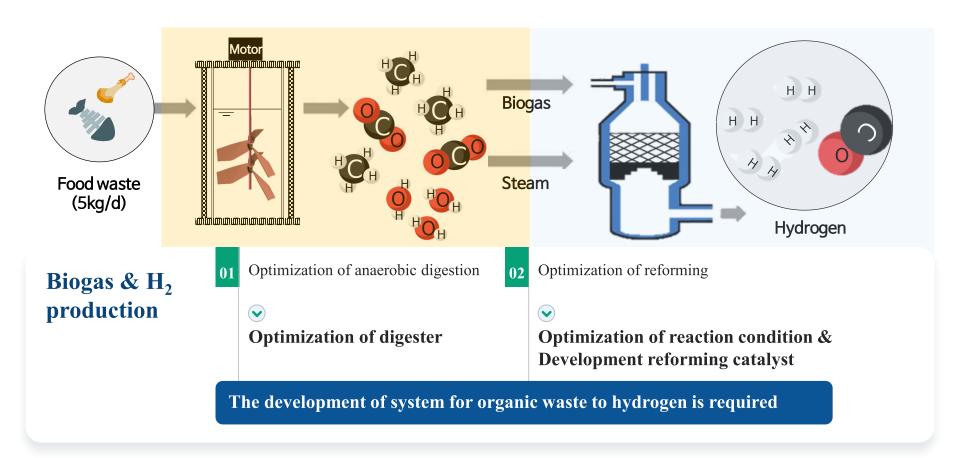
Hydrogen production from biogas generated from organic waste (Converting CH<sub>4</sub> and CO<sub>2</sub> into H<sub>2</sub> and CO through reforming of biogas with steam)

**Steam reforming of biogas** Instead of combustion of biogas to produce heat





# Research goal





Research 1. Anaerobic digestion

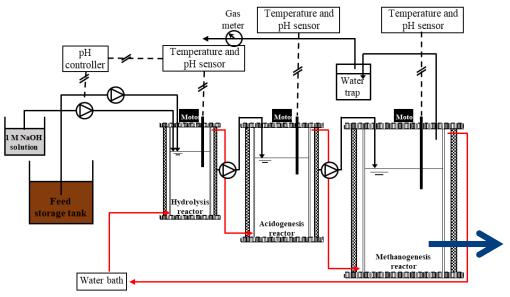
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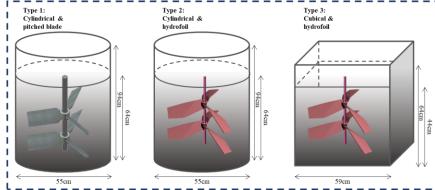


# **Optimization of Anaerobic digestion**

### \* Anaerobic digestor



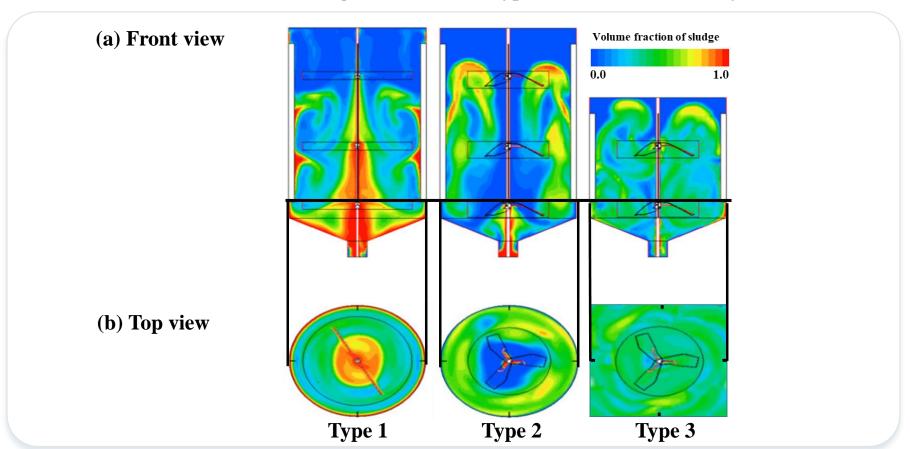
#### Type of reactor and agitator





# Simulation results (computational fluid dynamics)

Volume fractions of sludge with different type of reactor obtained by CFD



- > Highly dispersed volume fractions of sludge is resulting from better mixing efficiency
- > Type 3 reactor showed the highest distribution of volume fraction due to the axial flow caused by Hydrofoil agitator and enhanced turbulence flow in edges of cubical reactor.



### **Monitoring results**

#### Pollutant removal efficiencies and biogas production

Indicator		Type 1 Cylindrical shape Pitched blade	Type 2 Cylindrical shape Hydrofoil	Type 3 Cubical shape Hydrofoil
Removal efficiency	TCOD (%)	65.8	66.3	76.5
	TS (%)	62.1	62.6	72.2
	VS (%)	62.2	62.8	72.4
	Biogas yield (L/kg FW*)	72.3	72.9	84.1
Biogas production	Methane yield	42.1	44.3	51.3
	(L/kg FW)	42.1		
	Methane content (%)	58.2	60.8	61.3

- A comparison of types 1 and 2 showed that hydrofoil agitator exhibited a higher removal efficiency and biogas production due to the axial flow caused by the hydrofoil agitator
- A comparison of types 2 and 3 revealed that the cubical reactor enhanced the COD, TS, and VS removal efficiencies and biogas production, which resulted from the turbulence flow



Research 2. Reforming of biogas

2





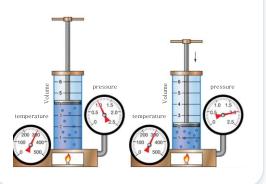
# **Optimization of reforming of methane**

#### **Parameters**

Reaction temperature

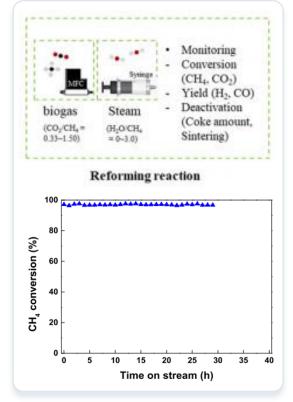
Oxidant/CH<sub>4</sub> ratio

Oxidant type



# **Simulation** Carbon formation Hydrogen production Minimum SEC Specific energy consumption (SEC) Maximum TE Thermal efficiency (TE)

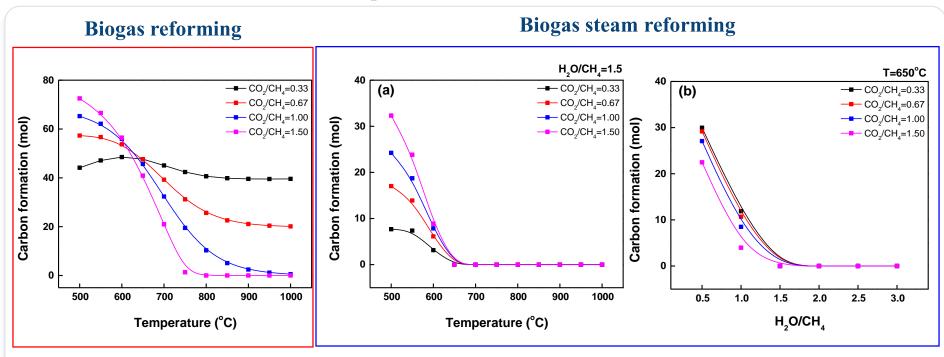
#### **Validation**





## Simulation results (optimization of oxidant)

#### The relationship between oxidant and carbon formation

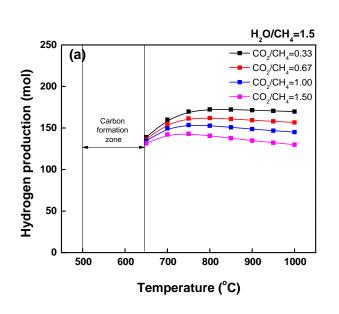


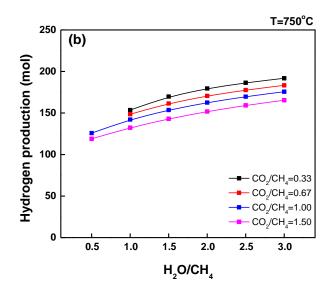
- > In biogas reforming, carbon formation can be suppressed by maintaining a reaction temperature of > 800 °C and a CO₂/CH₄ ratio of > 1.50.
- In biogas steam reforming,  $H_2O/CH_4$  ratio > 1.5 and temperature > 650 °C were required to prevent the carbon formation.



# Simulation results (Reaction conditions and H<sub>2</sub> production)

#### The relationship between reaction conditions and hydrogen production



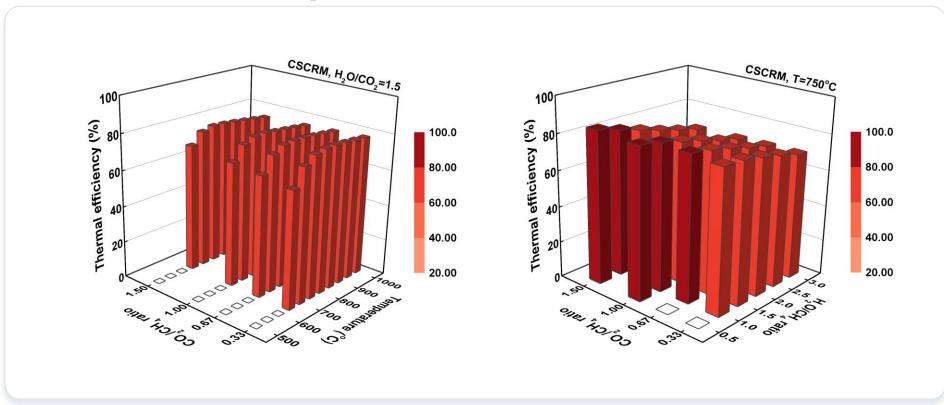


- >  $\rm H_2$  production increased up to 750 °C and decreased thereafter, regardless of the biogas composition
- ▶ Hydrogen production increase with increasing steam/CH<sub>4</sub> ratio and increasing CH<sub>4</sub> content in biogas



### **Simulation results** (Reaction conditions and H<sub>2</sub> production)

#### The relationship between reaction conditions and thermal efficiencies

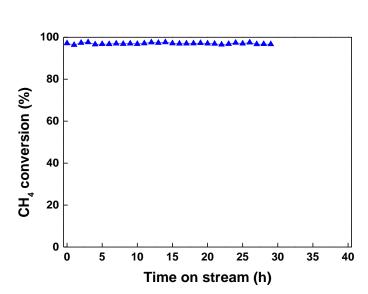


- > At > 800 °C, thermal efficiency decreased with increasing temperature; thus, 750 °C was the optimal temperature for the steam reforming of biogas.
- Although a high thermal efficiency was achieved at a low H<sub>2</sub>O/CH<sub>4</sub> ratio. However, low H<sub>2</sub>O/CH<sub>4</sub> ratio causes carbon formation, the optimum ratio was fixed at 1.5.



### **Validation results**

#### The catalyst activity test and deactivation study



Catal at	Coke	Ni crystallite size (nm)		
Catalyst	$(g/g_{cat})$	Reduced	Used	
Ni-MgO-CeZrO <sub>2</sub>	0	11.33	15.19 (34%)	

- > CH<sub>4</sub> conversion over the Ni-MgO-CeZrO<sub>2</sub> catalyst was maintained for 30 h without a significant decrease, indicating that the optimized reaction conditions were energy effective and favored stable H<sub>2</sub> production in the real biogas steam reforming reaction.
- > No carbon was deposited on the catalyst during biogas reforming under optimized conditions.
- > Ni crystallite size slightly increased but it was not affect to deactivation of catalyst.



**Future research plans** 







## Demonstration project for hydrogen production



# Scale up

Process scale up

- Catalyst production scale
- Plantscale

### Data acquisition

**Biogas data** 

하수처리시설

- -Composition
- Impurities

### Scenario study

Techno environ. Analysis

- Economic efficiency Environmental impact

